## Acta Crystallographica Section C

## Crystal Structure

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# Dysobinin, a tetranortriterpenoid ${ }^{1}$ 

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The planar furan ring in the title compound ( $6 \beta$-acetoxyazadirone, $\mathrm{C}_{30} \mathrm{H}_{38} \mathrm{O}_{6}$ ) is twisted with respect to the steroid $D$ ring. The crystal structure is stabilized by $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds and van der Waals interactions.

## Comment

Dysobinin is a tetranortriterpenoid belonging to the meliacin class of compounds and was isolated from Dysoxylum binecteriferum in around $2 \%$ yield (Singh et al., 1976). It is an example of a growing family of meliacins and is of chemotaxonomic importance as it is a 6-acetoxy derivative of azadirone occurring in Melia azadirachta, a plant of the same family. This, together with the structural diversity of meliacin, prompted us to undertake the present study of (I).

(I)

The molecule contains one fused-ring system $(A / B / C / D)$ with eight chiral centres and one furan ring $(E)$. The torsion angles (Table 1) and least-squares-plane calculations (Table 2) indicate that ring $A$ is a puckered sofa, while rings $B$ and $C$ adopt a distorted chair and a boat conformation, respectively. These three rings are trans-fused with each other. The cyclopentene ring $(D)$ is in an envelope conformation. The furan ring $(E)$ is planar, $\alpha$-substituted and twisted with respect to the cyclopentene ring.

The present study does not establish the absolute configuration of the title molecule. However, based on literature precedence (Lavie et al., 1971), all the triterpenoids have the methyl group at $\mathrm{C} 10 \beta$-oriented in their absolute configuration. Accordingly, the C18 and C19 methyl groups in the title molecule have a $\beta$ orientation, while the C 20 methyl group has an $\alpha$ orientation. The two acetoxy groups at the chiral centres, C 6 and C7, whose relative stereochemistry is $S$ and $R$, have $\alpha$ equatorial and $\alpha$-axial configurations, respectively. The crystal structure analysis reveals the presence of weak intermolecular $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ interactions (Table 3). These weak hydrogen-bond interactions play a significant role in the stabilization of the solid-state structure (Desiraju, 1996).

## Experimental

Dysobinin was isolated from the alcoholic extract of the air-dried powdered fruits of Dysoxylum binecteriferum (Singh et al., 1976). Diffraction-quality crystals were obtained by slow evaporation from a methanol solution at room temperature.

## Crystal data

$\mathrm{C}_{30} \mathrm{H}_{38} \mathrm{O}_{6}$
$M_{r}=494.60$
Orthorhombic, $P 2_{1} 2_{1} 2$
$a=12.426$ (3) A
$b=29.446$ (10) $\AA$
$c=7.450(9) \AA$
$V=2726(3) \AA^{3}$
$Z=4$
Mo $K \alpha$ radiation
Cell parameters from 25 reflections
$\theta=5.84-9.98^{\circ}$
$\mu=0.083 \mathrm{~mm}^{-1}$
$T=293$ (2) K
Block, colourless
$0.30 \times 0.25 \times 0.20 \mathrm{~mm}$
$D_{x}=1.205 \mathrm{Mg} \mathrm{m}^{-3}$
Data collection
Enraf-Nonius CAD-4 diffractometer
$\theta_{\text {max }}=24.94^{\circ}$
$h=0 \rightarrow 13$
$\omega-2 \theta$ scans
$k=0 \rightarrow 31$
4725 measured reflections
2658 independent reflections
1552 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.050$
3 standard reflections frequency: 60 min intensity decay: $<0.5 \%$

## Refinement

Refinement on $F^{2}$
H -atom parameters constrained
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.045$
$w=1 /\left[\sigma^{2}\left(F_{o}{ }^{2}\right)+(0.0716 P)^{2}\right]$
where $P=\left(F_{o}{ }^{2}+2 F_{c}{ }^{2}\right) / 3$
$S=0.971$
$(\Delta / \sigma)_{\max }=0.004$
2658 reflections
332 parameters
$\Delta \rho_{\text {max }}=0.21 \mathrm{e}^{-3}$
$\Delta \rho_{\min }=-0.22 \mathrm{e}_{\AA^{-3}}$

Table 1
Selected torsion angles ( ${ }^{\circ}$ ).

| $\mathrm{C} 10-\mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 3$ | $1.3(7)$ | $\mathrm{C} 6-\mathrm{C} 5-\mathrm{C} 10-\mathrm{C} 9$ | $56.2(4)$ |
| :--- | ---: | :--- | ---: |
| $\mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 3-\mathrm{C} 4$ | $-27.8(7)$ | $\mathrm{C} 8-\mathrm{C} 9-\mathrm{C} 10-\mathrm{C} 5$ | $-54.0(4)$ |
| $\mathrm{C} 2-\mathrm{C} 3-\mathrm{C} 4-\mathrm{C} 5$ | $8.4(5)$ | $\mathrm{C} 8-\mathrm{C} 9-\mathrm{C} 11-\mathrm{C} 12$ | $39.7(6)$ |
| $\mathrm{C} 3-\mathrm{C} 4-\mathrm{C} 5-\mathrm{C} 10$ | $33.5(5)$ | $\mathrm{C} 9-\mathrm{C} 11-\mathrm{C} 12-\mathrm{C} 13$ | $16.1(7)$ |
| $\mathrm{C} 10-\mathrm{C} 5-\mathrm{C} 6-\mathrm{C} 7$ | $-60.3(4)$ | $\mathrm{C} 11-\mathrm{C} 12-\mathrm{C} 13-\mathrm{C} 20$ | $72.3(6)$ |
| $\mathrm{H} 5-\mathrm{C} 5-\mathrm{C} 10-\mathrm{C} 18$ | 174.8 | $\mathrm{C} 11-\mathrm{C} 12-\mathrm{C} 13-\mathrm{C} 14$ | $-52.6(6)$ |
| $\mathrm{C} 19-\mathrm{C} 8-\mathrm{C} 9-\mathrm{H} 9$ | 175.6 | $\mathrm{C} 9-\mathrm{C} 8-\mathrm{C} 14-\mathrm{C} 13$ | $17.7(5)$ |
| $\mathrm{O} 2-\mathrm{C} 6-\mathrm{C} 7-\mathrm{O} 4$ | $61.0(4)$ | $\mathrm{C} 17-\mathrm{C} 13-\mathrm{C} 14-\mathrm{C} 15$ | $-20.4(5)$ |
| $\mathrm{C} 5-\mathrm{C} 6-\mathrm{C} 7-\mathrm{C} 8$ | $58.2(4)$ | $\mathrm{C} 12-\mathrm{C} 13-\mathrm{C} 14-\mathrm{C} 8$ | $35.1(5)$ |
| C6-C7-C8-C9 | $-51.0(4)$ | $\mathrm{C} 13-\mathrm{C} 14-\mathrm{C} 15-\mathrm{C} 16$ | $1.9(6)$ |
| C14-C8-C9-C11 | $-55.8(4)$ | $\mathrm{C} 14-\mathrm{C} 15-\mathrm{C} 16-\mathrm{C} 17$ | $18.2(6)$ |
| C7-C8-C9-C10 | $50.8(4)$ | $\mathrm{C} 15-\mathrm{C} 16-\mathrm{C} 17-\mathrm{C} 13$ | $-29.5(5)$ |
| C2-C1-C10-C5 | $39.2(5)$ | $\mathrm{C} 14-\mathrm{C} 13-\mathrm{C} 17-\mathrm{C} 16$ | $30.3(4)$ |
| C4-C5-C10-C1 | $-55.6(4)$ | $\mathrm{C} 16-\mathrm{C} 17-\mathrm{C} 30-\mathrm{C} 29$ | $33.2(7)$ |

[^0]Table 2
Least-squares-planes data showing the deviations of atoms from the mean plane defined by atoms marked with an asterisk (*).

| Plane 1 |  | Plane 2 |  |
| :---: | :---: | :---: | :---: |
| C1* | 0.1239 | C5* | -0.0301 |
| C2* | -0.1614 | C6* | 0.0300 |
| C3* | 0.0742 | C7 | -0.6321 |
| C4* | 0.0322 | C8* | -0.0296 |
| C5* | -0.0689 | C9* | 0.0297 |
| C10 | 0.6730 | C10 | 0.6852 |
| C18 | 2.2072 | C18 | 2.2436 |
|  |  | C19 | 1.3300 |
| Plane 3 |  |  |  |
| C8* | 0.1045 | Plane 4 |  |
| C9 | -0.5947 | C13* | -0.0063 |
| C11* | -0.1043 | C14* | 0.0119 |
| C12* | 0.1070 | C15* | -0.0121 |
| C13 | -0.5449 | C16* | 0.0065 |
| C14* | -0.1073 | C17 | 0.5085 |
| C19 | 1.6297 |  |  |
| C20 | -2.0582 |  |  |
|  |  | Plane 6 |  |
|  |  | C27* | 0.0030 |
| Plane 5 |  | C28* | 0.0012 |
| C13* | -0.1711 | C29* | 0.0005 |
| C14* | 0.0783 | C30* | -0.0021 |
| C15* | 0.0499 | O6* | -0.0026 |
| C16* | -0.1613 |  |  |
| C17* | 0.2043 |  |  |
| C30 | -0.3501 |  |  |

Data collection: CAD-4-MACH/PC (Enraf-Nonius, 1993); cell refinement: $C A D-4-M A C H / P C$; data reduction: $N R C V A X$ (Gabe et al., 1989); program(s) used to solve structure: SHELXS97 (Sheldrick, 1990); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: $N R C V A X$; software used to prepare material for publication: SHELXL97.

Table 3
Hydrogen-bonding geometry $\left(\AA^{\circ}{ }^{\circ}\right)$.

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{C} 9-\mathrm{H} 9 \cdots \mathrm{OB}^{\mathrm{i}}$ | 0.98 | 2.57 | $3.437(6)$ | 147 |
| $\mathrm{C} 20-\mathrm{H} 20 C \cdots \mathrm{OB}^{\mathrm{i}}$ | 0.96 | 2.57 | $3.501(6)$ | 164 |
| $\mathrm{C} 28-\mathrm{H} 28 \cdots \mathrm{O} 5^{\mathrm{ii}}$ | 0.93 | 2.54 | $3.413(8)$ | 157 |
| Symmetry codes: (i) $x, y, z-1 ;$ (ii) $2-x,-y, z-1$. |  |  |  |  |

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[^0]:    ${ }^{1}$ CDRI communication No. 5802.

